

PFAS-Free Industrial Innovation: Developing Sustainable Alternatives for Chemical and Manufacturing Applications

Ted Baker

Professor Rutgers University

Abstract

Per- and polyfluoroalkyl substances (PFAS) have been extensively used in industrial applications because of their exceptional chemical stability, thermal resistance, low surface energy, and resistance to water, oil, and stains. However, the environmental persistence, widespread distribution, and potential health concerns associated with many PFAS have accelerated regulatory pressure and industrial interest in safer alternatives. This paper examines emerging pathways for PFAS-free industrial innovation across chemical processing, manufacturing, textiles, food packaging, electronics, construction, automotive, aerospace, and surface-treatment applications. A conceptual qualitative methodology is adopted to analyse alternative material classes, substitution strategies, performance requirements, circular-economy considerations, and implementation challenges. The study evaluates potential alternatives including silicones, hydrocarbons, polyolefins, bio-based polymers, mineral coatings, sol-gel systems, waxes, polyurethane-based formulations, and advanced hybrid materials. Particular attention is given to the principle that PFAS substitution should not involve simple chemical replacement but should consider complete life-cycle performance, toxicity, persistence, recyclability, energy consumption, and end-of-life management. The paper proposes an Integrated PFAS-Free Industrial Innovation Framework based on hazard assessment, functional redesign, material substitution, performance validation, life-cycle assessment, and circular manufacturing. The analysis suggests that successful transition will depend on application-specific material engineering rather than a single universal replacement. Digital materials discovery, artificial intelligence, green chemistry, surface engineering, and scalable manufacturing could accelerate the development of high-performance PFAS-free solutions. The paper concludes that PFAS-free industrial innovation can become an important component of sustainable manufacturing when environmental safety, technical performance, economic feasibility, and circularity are addressed simultaneously.

Keywords: PFAS, PFAS-Free Materials, Sustainable Manufacturing, Green Chemistry, Chemical Substitution, Industrial Innovation, Sustainable Materials, Surface Engineering, Circular Economy, Safer Chemicals.

1. Introduction

Per- and polyfluoroalkyl substances, commonly referred to as PFAS, comprise a large family of fluorinated chemicals that have been used in numerous industrial and consumer applications.

Their industrial value is associated with properties such as:

- High chemical stability.
- Thermal resistance.
- Low surface energy.
- Oil and water repellence.
- Non-stick behaviour.
- Resistance to degradation.

These characteristics have made PFAS useful in applications ranging from specialised coatings to industrial processing and high-performance products.

However, many PFAS are highly persistent in the environment. Certain well-studied PFAS have also been associated with adverse health and environmental concerns.

Consequently, governments, manufacturers, researchers, and downstream users are increasingly investigating PFAS-free alternatives.

The challenge is substantial because PFAS are often used precisely where conventional materials struggle to provide the same combination of performance characteristics.

2. The PFAS Challenge

The central industrial challenge can be expressed as:

High Performance + Chemical Safety + Environmental Persistence + Economic Feasibility

Historically, industrial material selection often prioritised performance and cost.

The emerging model increasingly requires consideration of:

Performance + Safety + Sustainability + Circularity

This shift is changing chemical and materials innovation.

3. Industrial Applications of PFAS

PFAS have been used in areas including:

- Textiles.
- Food-contact materials.
- Firefighting applications.

- Electronics.
- Semiconductor manufacturing.
- Automotive components.
- Aerospace systems.
- Industrial coatings.
- Lubrication.
- Metal processing.
- Construction materials.
- Medical technologies.

The exact PFAS used and their regulatory status vary considerably by application.

4. Research Objectives

This paper aims to:

1. Examine the industrial importance of PFAS.
2. Analyse drivers of PFAS substitution.
3. Identify emerging PFAS-free material alternatives.
4. Examine green-chemistry approaches to substitution.
5. Evaluate application-specific alternatives.
6. Analyse performance and durability challenges.
7. Examine life-cycle implications.
8. Explore circular manufacturing opportunities.
9. Examine AI-enabled materials discovery.
10. Propose an integrated PFAS-free innovation framework.

5. Research Methodology

A conceptual qualitative research approach is adopted.

The analysis considers developments across:

- Materials science.
- Polymer chemistry.
- Surface engineering.
- Green chemistry.
- Industrial manufacturing.
- Environmental management.
- Circular economy.

Alternative technologies are evaluated according to:

Criterion	Key Question
Functional performance	Can the alternative perform the required function?
Toxicity	Is the replacement demonstrably safer?
Persistence	Does it avoid long-term environmental persistence?
Durability	Can it survive the intended service conditions?
Cost	Is commercial adoption feasible?
Scalability	Can production be expanded?
Recyclability	Can the material enter circular systems?
Life-cycle impact	What are the total environmental consequences?

6. Why PFAS Are Difficult to Replace

PFAS possess an unusual combination of properties.

Their fluorinated structures can provide:

- Strong chemical resistance.
- Low friction.
- Thermal stability.
- Hydrophobicity.
- Oleophobicity.

Replacing one property is relatively straightforward.

Replacing the complete performance package is considerably more difficult.

7. From Chemical Substitution to Functional Redesign

A major principle of sustainable substitution is:

Do not simply replace one molecule with another; redesign the function.

For example, instead of asking:

"What chemical can replace this PFAS?"

industrial researchers can ask:

"How can the product achieve the required performance without requiring fluorinated chemistry?"

This opens opportunities for:

- New surface structures.
- Different manufacturing processes.

- Mechanical redesign.
- Alternative coatings.
- Material combinations.

8. Green Chemistry

Green chemistry provides an important foundation for PFAS-free innovation.

Relevant principles include:

- Prevention of waste.
- Safer chemical synthesis.
- Reduced toxicity.
- Renewable feedstocks.
- Energy efficiency.
- Degradation after use.
- Reduced environmental persistence.

The ideal alternative should not simply eliminate PFAS while introducing another persistent or hazardous substance.

9. Silicone-Based Alternatives

Silicone chemistry provides useful properties including:

- Water repellence.
- Thermal stability.
- Flexibility.
- Surface modification.

Silicone-based coatings may therefore serve as alternatives in selected applications.

However, their environmental performance must be assessed case by case.

10. Hydrocarbon-Based Materials

Hydrocarbon polymers and wax-based systems can provide:

- Water resistance.
- Surface protection.
- Lubrication.
- Barrier properties.

Examples include:

- Polyethylene.
- Polypropylene.
- Hydrocarbon waxes.

Their suitability depends strongly on the intended application.

11. Polyolefin-Based Alternatives

Polyolefins offer:

- Low cost.
- Chemical resistance.
- Manufacturing scalability.
- Established recycling infrastructure in some markets.

They may be suitable for selected packaging and coating applications.

12. Bio-Based Polymers

Renewable feedstocks can support development of:

- Bio-based polyesters.
- Cellulose derivatives.
- Starch-based materials.
- Natural polymer coatings.
- Plant-derived surface treatments.

Bio-based origin alone does not guarantee environmental superiority; complete life-cycle assessment remains necessary.

13. Cellulose-Based Barrier Materials

Cellulose can be engineered to provide:

- Oxygen barriers.
- Grease resistance.
- Structural strength.

Advanced cellulose nanomaterials may support PFAS-free packaging applications.

14. Mineral-Based Coatings

Mineral systems can provide functional surface properties through:

- Silica.
- Alumina.
- Calcium-based materials.
- Clay minerals.

These systems can potentially replace fluorinated surface treatments in selected applications.

15. Sol-Gel Technologies

Sol-gel chemistry enables thin inorganic or hybrid coatings.

Potential benefits include:

- Surface protection.
- Controlled wettability.
- Thermal resistance.
- Chemical resistance.

Sol-gel coatings are particularly interesting for industrial surface engineering.

16. Hybrid Organic–Inorganic Materials

Combining organic polymers with inorganic structures can produce multifunctional coatings.

Potential properties include:

- Abrasion resistance.
- Barrier performance.
- Hydrophobicity.
- Thermal stability.

Hybrid systems may therefore provide an alternative route to fluorine-free surface engineering.

17. Textile Applications

PFAS have historically been used to provide water and oil repellence to textiles.

Potential PFAS-free alternatives include:

- Silicone treatments.
- Hydrocarbon-based finishes.
- Dendrimer-inspired coatings.
- Wax-based treatments.
- Bio-based surface treatments.

Performance must be assessed for:

- Water repellence.
- Oil repellence.
- Washing durability.
- Breathability.
- Fabric feel.

18. Food Packaging

Food packaging represents an important substitution area.

PFAS have been used in some applications because of their grease and oil resistance.

Potential alternatives include:

- Cellulose-based coatings.
- Mineral coatings.
- Biopolymer barriers.
- Polyolefin systems.
- Water-based dispersions.

Food-contact safety must be independently evaluated for each formulation.

19. Paper and Board Packaging

PFAS-free barrier technologies can combine:

Fibre Structure + Mineral Barrier + Bio-Based Polymer

Such systems may provide grease and moisture resistance without relying on fluorinated chemistry.

20. Industrial Coatings

Industrial coatings require combinations of:

- Adhesion.
- Durability.
- Chemical resistance.
- Corrosion protection.
- Low friction.

PFAS-free solutions may use:

- Silicones.
- Polyurethanes.
- Epoxy systems.
- Sol-gel coatings.
- Ceramic surfaces.

21. Non-Stick Applications

PFAS-containing fluoropolymers have historically been important for non-stick properties.

Alternative strategies can include:

- Ceramic coatings.
- Sol-gel systems.
- Surface texturing.
- Silicon-based coatings.
- Advanced polymer composites.

Performance requirements remain application-specific.

22. Lubrication

Fluorinated lubricants provide exceptional performance under demanding conditions.

Potential alternatives include:

- Synthetic hydrocarbons.
- Esters.
- Silicones.
- Bio-based lubricants.
- Solid lubricants.

Selection should consider temperature, pressure, oxidation, wear, and chemical compatibility.

23. Electronics

PFAS-related materials may appear in specialised electronics manufacturing and processing.

Potential PFAS-free innovation areas include:

- Alternative cleaning systems.
- Non-fluorinated coatings.
- Modified photoresist chemistries.
- Alternative process materials.

However, semiconductor manufacturing is particularly demanding because contamination control and nanoscale precision are critical.

24. Semiconductor Manufacturing

Advanced semiconductor manufacturing requires:

- High purity.
- Thermal stability.
- Chemical compatibility.
- Extreme process control.

PFAS substitution therefore represents a technically challenging research area.

Potential pathways include:

- New polymer chemistries.
- Alternative process fluids.
- Surface-engineered materials.
- Closed-loop recovery systems.

25. Automotive Applications

Potential applications include:

- Wire and cable coatings.
- Gaskets.
- Seals.
- Lubricants.
- Interior textiles.
- Surface treatments.

PFAS-free alternatives must withstand:

- Temperature.
- Vibration.
- Mechanical wear.
- Chemical exposure.

26. Aerospace

Aerospace materials require exceptionally high reliability.

Potential PFAS-free innovation areas include:

- Coatings.
- Lubricants.

- Sealants.
- Textiles.
- Composite processing.

Qualification requirements are significantly more demanding than ordinary consumer applications.

27. Firefighting Applications

Aqueous film-forming foams have historically relied on fluorinated chemistry for certain high-performance fire suppression applications.

PFAS-free firefighting foams are an important area of innovation.

Alternatives must demonstrate:

- Fire suppression effectiveness.
- Storage stability.
- Compatibility with equipment.
- Environmental safety.

28. Construction Materials

PFAS-free materials can be developed for:

- Roofing.
- Waterproofing.
- Sealants.
- Protective coatings.
- Insulation systems.

Construction applications offer significant opportunities because large material volumes are deployed over long periods.

29. Water and Oil Repellence

One of the most difficult functions to reproduce without fluorinated chemistry is simultaneous:

Water Repellence + Oil Repellence

Alternative approaches include:

- Surface texturing.
- Silicone chemistry.
- Hydrocarbon structures.
- Nanostructured surfaces.

30. Surface Engineering

Instead of changing the entire material, engineers can modify only the surface.

Possible techniques include:

- Plasma treatment.
- Laser texturing.
- Micro-patterning.
- Nanostructuring.
- Sol-gel coating.

This can reduce material consumption.

31. Superhydrophobic Surfaces

Superhydrophobicity can be produced through a combination of:

Low Surface Energy + Surface Roughness

Biomimetic surface structures inspired by lotus leaves demonstrate how physical structure can contribute to water repellence.

32. Biomimicry

Nature provides examples of:

- Water-repellent leaves.
- Self-cleaning surfaces.
- Low-friction biological structures.
- Protective coatings.

Engineering these principles could reduce reliance on persistent fluorinated chemistry.

33. Micro- and Nanostructured Surfaces

Surface topography can change:

- Wettability.
- Friction.
- Adhesion.
- Contamination behaviour.

This creates opportunities to replace chemical functionality with physical functionality.

34. Advanced Polymer Design

Polymer chemistry can be redesigned to provide:

- Barrier performance.
- Chemical resistance.
- Flexibility.
- Durability.

Computational modelling can accelerate polymer discovery.

35. Artificial Intelligence for PFAS-Free Materials

AI can support:

- Molecular discovery.
- Polymer-property prediction.
- Formulation optimisation.
- Toxicity screening.
- Life-cycle prediction.

A conceptual workflow is:

Material Database → AI Screening → Candidate Selection → Laboratory Validation → Scale-Up

36. Machine Learning and Materials Discovery

Machine-learning models can identify relationships between:

- Molecular structure.
- Surface energy.
- Thermal properties.
- Mechanical performance.
- Chemical stability.

This can reduce the number of experimental formulations required.

37. Digital Twins for Manufacturing

Digital twins can model manufacturing systems before physical deployment.

They can evaluate:

- Material consumption.
- Energy demand.
- Process efficiency.
- Waste generation.
- Product performance.

This can help optimise PFAS-free production processes.

38. Circular Economy

PFAS-free innovation should be connected to circular manufacturing.

A circular model includes:

Design → Production → Use → Recovery → Recycling → Reuse

Materials should be designed with end-of-life management in mind.

39. Design for Recycling

A PFAS-free product is not automatically recyclable.

Additional considerations include:

- Material compatibility.
- Additive selection.

- Adhesives.
- Coatings.
- Composite complexity.

Design simplification can improve recyclability.

40. Life-Cycle Assessment

PFAS alternatives should be evaluated using life-cycle assessment.

The assessment should cover:

1. Raw materials.
2. Manufacturing.
3. Transport.
4. Product use.
5. Maintenance.
6. Recycling.
7. Disposal.

41. Avoiding Regrettable Substitution

A critical principle is avoiding regrettable substitution.

Replacing PFAS with another poorly understood persistent or hazardous chemical would not represent genuine sustainability.

Therefore:

PFAS-Free ≠ Automatically Sustainable

The replacement must be assessed independently.

42. Performance Validation

PFAS-free alternatives should undergo rigorous testing.

Key indicators may include:

- Durability.
- Thermal resistance.
- Chemical resistance.
- Water repellence.
- Oil resistance.
- Mechanical strength.
- Ageing behaviour.

43. Economic Feasibility

Industrial adoption depends heavily on cost.

Important factors include:

- Raw material prices.

- Equipment modification.
- Process changes.
- Labour requirements.
- Certification.
- Maintenance.
- Product lifetime.

44. Supply-Chain Considerations

Companies need reliable access to alternative materials.

PFAS substitution may therefore require:

- New suppliers.
- New manufacturing equipment.
- Reformulation.
- Quality-control systems.
- New certifications.

45. Regulatory Drivers

Growing regulatory attention toward PFAS is accelerating substitution research.

Companies increasingly need to understand:

- Chemical restrictions.
- Product requirements.
- Reporting obligations.
- Supply-chain transparency.

Regulatory requirements vary between jurisdictions and applications, so compliance should be assessed using current applicable rules.

46. Corporate Sustainability

PFAS-free innovation can contribute to:

- Safer product design.
- Chemical-risk reduction.
- Environmental stewardship.
- Sustainable procurement.
- Corporate responsibility.

However, sustainability claims should be supported by evidence rather than based solely on the absence of PFAS.

47. Integrated PFAS-Free Innovation Framework

The proposed framework contains seven stages:

Stage 1 — Functional Identification

Determine why PFAS is used.

Stage 2 — Hazard Assessment

Evaluate potential replacement chemistries.

Stage 3 — Functional Redesign

Investigate whether the function can be achieved differently.

Stage 4 — Material Development

Develop alternative materials or coatings.

Stage 5 — Performance Testing

Validate technical requirements.

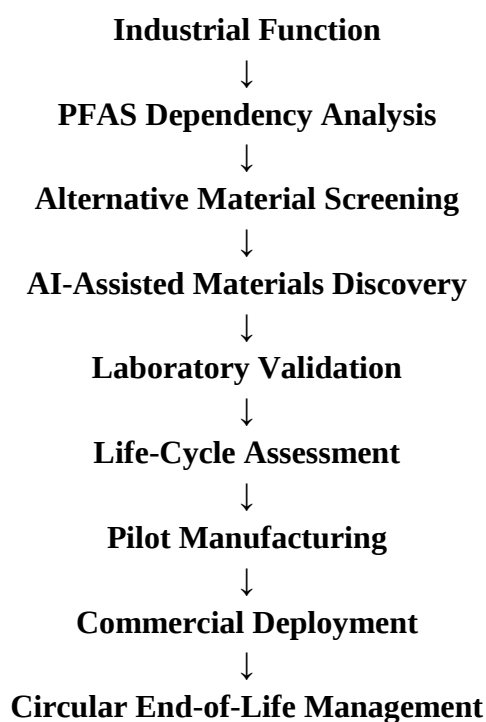
Stage 6 — Life-Cycle Assessment

Evaluate environmental consequences.

Stage 7 — Industrial Scale-Up

Optimise manufacturing and supply chains.

48. Framework Architecture



49. Comparative Assessment

Alternative	Potential Advantages	Key Challenges
Silicone	Water repellence, flexibility	Performance varies by application
Hydrocarbon polymers	Low cost, scalability	Limited high-temperature performance
Bio-based polymers	Renewable feedstocks	Durability and moisture sensitivity
Sol-gel coatings	Surface functionality	Processing complexity
Mineral coatings	Thermal and barrier properties	Brittleness in some systems
Polyolefins	Established processing	Limited specialised functionality
Ceramic coatings	High temperature resistance	Processing and cracking
Nanostructured surfaces	Reduced chemical dependence	Scale-up challenges

50. Industrial Adoption Strategy

Organisations should begin with:

High-Volume + High-Exposure + Technically Feasible applications.

A staged approach reduces technical and financial risks.

51. Role of Research and Development

Universities and industry should collaborate on:

- New polymer chemistry.
- Surface engineering.
- Green synthesis.
- Materials characterisation.
- Toxicological assessment.
- Pilot-scale manufacturing.

Cross-disciplinary collaboration is particularly important.

52. Standardisation

The industry needs reliable methods for comparing PFAS-free alternatives.

Testing protocols should address:

- Performance.
- Environmental impact.
- Durability.
- Chemical safety.

- End-of-life behaviour.

53. Challenges

Major challenges include:

1. Performance gaps.
2. High development costs.
3. Lack of long-term field data.
4. Manufacturing compatibility.
5. Alternative chemical uncertainty.
6. Recycling complexity.
7. Regulatory differences.
8. Consumer expectations.
9. Supply-chain transformation.
10. Certification requirements.

54. Future Research Directions

Future research should focus on:

- PFAS-free high-performance coatings.
- Bio-based repellents.
- Fluorine-free fire suppression.
- Advanced barrier packaging.
- PFAS-free semiconductor materials.
- Non-fluorinated lubricants.
- AI-assisted chemical discovery.
- Recyclable surface treatments.
- Low-energy manufacturing.
- Safer-by-design materials.

55. Discussion

PFAS substitution represents more than a chemical replacement challenge.

It is fundamentally a systems innovation problem.

In many cases, industries have historically relied on PFAS because the materials provide multiple functions simultaneously.

Consequently, replacing PFAS may require a combination of:

New Chemistry + Surface Engineering + Product Redesign + Process Innovation

For example, water resistance that previously depended on fluorinated chemistry might instead be achieved through a combination of material composition and surface microstructure.

This demonstrates why functional redesign can be more powerful than direct chemical substitution.

56. Strategic Recommendations

Industrial organisations should:

1. Map PFAS use across their supply chains.
2. Identify the exact function provided by each PFAS-containing material.
3. Prioritise high-risk applications.
4. Develop multiple alternative technologies.
5. Use life-cycle assessment.
6. Avoid regrettable substitutions.
7. Establish long-term performance testing.
8. Invest in green chemistry.
9. Use AI for materials screening.
10. Engage suppliers and customers during transition.

57. Conclusion

PFAS-free industrial innovation represents an important transition in sustainable chemical and manufacturing systems.

The challenge is not simply to remove fluorinated substances but to develop alternatives that simultaneously provide:

Performance + Safety + Durability + Affordability + Circularity

Silicone systems, hydrocarbons, bio-based polymers, mineral coatings, sol-gel technologies, advanced ceramics, polyolefins, and engineered surface structures provide promising pathways for selected applications.

However, no universal replacement exists.

The most effective strategy is therefore application-specific and based on functional redesign. Emerging technologies such as artificial intelligence, computational materials science, nanostructured surfaces, digital twins, and green chemistry can accelerate this transition.

Ultimately, successful PFAS-free manufacturing will depend on integrating materials innovation with life-cycle thinking, safer-by-design principles, responsible regulation, and circular-economy strategies.

References

1. Buck, R. C., Franklin, J., Berger, U., et al. (2011). Perfluoroalkyl and polyfluoroalkyl substances in the environment: Terminology, classification, and origins. *Integrated Environmental Assessment and Management*, 7(4), 513–541.

2. Wang, Z., DeWitt, J. C., Higgins, C. P., & Cousins, I. T. (2017). A never-ending story of per- and polyfluoroalkyl substances (PFASs)? *Environmental Science & Technology*, 51(5), 2508–2518.
3. Glüge, J., Scheringer, M., Cousins, I. T., et al. (2020). An overview of the uses of per- and polyfluoroalkyl substances (PFAS). *Environmental Science: Processes & Impacts*, 22, 2345–2373.
4. Sunderland, E. M., Hu, X. C., Dassuncao, C., et al. (2019). A review of the pathways of human exposure to poly- and perfluoroalkyl substances (PFASs) and present understanding of health effects. *Journal of Exposure Science & Environmental Epidemiology*, 29, 131–147.
5. Cousins, I. T., DeWitt, J. C., Glüge, J., et al. (2022). Strategies for reducing the global burden of PFAS. *Environmental Science & Technology*, 56, 11139–11148.
6. Kwiatkowski, C. F., Andrews, D. Q., Birnbaum, L. S., et al. (2020). Scientific basis for managing PFAS as a chemical class. *Environmental Science & Technology Letters*, 7(8), 532–543.
7. Wang, Z., Boucher, J. M., Scheringer, M., Cousins, I. T., & Hungerbühler, K. (2017). Toward a comprehensive global emission inventory of C4–C10 perfluoroalkanesulfonic acids. *Environmental Science & Technology*, 51(8), 4481–4490.
8. Ameduri, B. (2018). Fluoropolymers: The right material for the right applications. *Chemical Communications*, 54, 8210–8219.
9. Schellenberger, S., Gillgard, P., Stare, A., et al. (2019). Facing the rain after the storm: Performance evaluation of alternative fluorine-free durable water repellents for outdoor apparel. *Chemosphere*, 220, 211–220.
10. Herzke, D., Schürmann, A., Nygård, T., et al. (2017). PFASs in bird eggs from remote environments. *Environmental Science & Technology*, 51, 10163–10172.
11. Lohmann, R., Cousins, I. T., DeWitt, J. C., et al. (2020). Are fluoropolymers really of low concern for human and environmental health and separate from other PFAS? *Environmental Science & Technology*, 54, 12820–12828.
12. Anastas, P. T., & Warner, J. C. (1998). *Green Chemistry: Theory and Practice*. Oxford University Press.
13. Zimmerman, J. B., Anastas, P. T., Erythropel, H. C., & Leitner, W. (2020). Designing for a green chemistry future. *Science*, 367(6476), 397–400.
14. Anastas, N. D., & Eghbali, N. (2010). *Green chemistry: Principles and case studies*. Chemical Society Reviews, 39, 301–312.
15. Geissdoerfer, M., Savaget, P., Bocken, N. M. P., & Hultink, E. J. (2017). The circular economy—A new sustainability paradigm? *Journal of Cleaner Production*, 143, 757–768.